SYNTHESIS OF 9-AZABICYCLONONANES FROM (Z.Z)-CYCLOOCTA-1.5-DIENE 1

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Summary: The transannular N-heterocyclization of cycloocta-1,5-diene with NBS in the presence of cyanamide is a preparative useful method for the synthesis of 9-azabicyclononanes; a mixture of endo,endo-2,5-dibromo-N-cyano-9-azabicyclo[4.2.1]nonane and endo,endo-2,6-dibromo-N-cyano-9-azabicyclo[3.3.1]nonane is obtained in high yield.

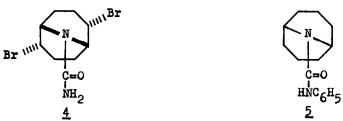
In the course of experiments designed to investigate the transannular reactivity of medium sized dienes 1,2 , we got interested in the reaction of (2,2)-cycloocta-1,5-diene $(\underline{1})$ with N-bromosuccinimide (NBS) in the presence of amines. But the reactions with primary aliphatic or aromatic amines were not succesful in obtaining any product of a transannular cyclization with participation of an amino group.

The formation of substituted 9-azabicyclononanes from cyclooctadiene is possible using mercury or thallium salts and primary arylamines $^{3-5}$ or N,N-dibromo-p-toluenesulfonamide in chloroform 6 .

Now we found that the known 7 reaction of alkenes with cyanamide and NBS in the case of 1 (molar ratio diene/NBS 1:2) in ether at room temperature occurs with transannular participation of the initially added cyanamido group.

A white crystalline 50:50 mixture (80% isolated yield) of endo,endo-2,5-di-bromo-N-cyano-9-azabicyclo[4.2.1]nonane (2) and endo,endo-2,6-dibromo-N-cyano-9-azabicyclo[3.3.1]nonane (3) 8,9 was obtained. Under the reaction conditions the product ratio is constant for minimal 72 h. Compound 2 (colorless needles, m.p. 107-108 °C) could be separated by several crystallizations from n-hexane, and compound 3 (colorless, fine needles, m.p. 138,5-139,5 °C) by several crystallizations of the residue from ethanol.

The stirring of the mixture of 2 and 3 (50:50) in 50% ag. sulfuric acid for 8 h at r.t. gave only one product (92% yield), the urea 4 8 (m.p. 198-200 °C). In addition to the hydrolysis of the cyanamido group under these acidic conditions an isomerization of the 9-azabicyclo[4.2.1]nonane system to the more stable 9-azabicyclo[3.3.1]nonane system took place.



The reduction of the mixture of $\underline{2}$ and $\underline{3}$ with LiAlH₄ in ether leads to an air sensitive ¹⁰ product, which could be trapped by treatment of the dried etheral solution with phenylisocyanate. An uniform product was isolated (87% yield, m.p. 193-194 °C) which was identified as the phenylurea $\underline{5}$ 8. Thus, the isomerization of the 9-azabicyclo[4.2.1]nonane system also occurs in a basic medium.

References and Notes

- Transannular Reactions of Cycloalkenes, Cycloalkadienes, and Cycloalkatrienes, Part IX. For part VIII. see G. Haufe, and M. Mühlstädt, <u>J. Prakt. Chem.</u> 323, 89 (1981).
- 2. G. Haufe, M. Mühlstädt, E. Kleinpeter, and J. Graefe, Monatsh. Chem. 109, 575 (1978).
- V. Gómez Aranda, J. Barluenga, G. Asensio, and M. Yus, <u>Tetrahedron Letters</u> 1972, 3621; V. Gómez Aranda, J. Barluenga, A.Ara, and G. Asensio, <u>Synthesis</u> 1974, 135; M. Barrelle, and M. Apparu <u>Tetrahedron</u> 33, 1309 (1977).
- 4. V. Gómez Aranda, J. Barluenga, and F. Aznar, Synthesis 1974, 504.
- 5. J. Barluenga, C. Jiménez, C. Najera, and M. Yus, Chem. Commun. 1981, 1178.
- 6. H. Stetter, and K. Heckel, Tetrahedron Letters 1972, 801.
- 7. K. Ponsold, and W. Ihn, Tetrahedron Letters 1970, 1125.
- 8. Satisfactory elemental analyses, MS, IR, and H-1-NMR spectra were obtained for all new compounds. Carbon-13-NMR spectra (22,63 MHz, TMS, CDCl₃):

 2: δ114,0 (CN), 65,2 (CHN), 51,1 (CHBr), 32,2 (CH₂CHBr), 27,1 (CH₂CHN);

 3: δ115,3 (CN), 55,6 (CHN), 47,8 (CHBr), 31,3 (CH₂CHBr), 25,3 (CH₂CHN);

 4 (DMSO): δ156,4 (CO), 52,3 (CHN), 49,6 (CHBr), 31,4 (CH₂CHBr), 25,1(CH₂CHN);

 5: δ171,1 (CO), 150,4, 128,8, 122,6, 119,8 (aryl carbons, α, m, p, ο),

 47,2 (CHN), 30,0 (CH₂CHN), 20,6 (CH₂-CH₂-CH₂).
- 9. For C-13-chemical shifts of 9-azabicyclo[4.2.1]nonane and 9-azabicyclo[3.3.1]nonane systems cf. A. G. Anastassiou, and E. Reichmanis, J. Amer.
 Chem. Soc. 98, 8267 (1976); M. Barrelle, M. Apparu, and C. Gey, Canad. J.
 Chem. 56, 85 (1978).
- 10. cf. A. G. Anastassiou, and R. P. Cellura, J. Org. Chem. 37, 3126 (1972).